

Magnesium based materials for hydrogen storage applications

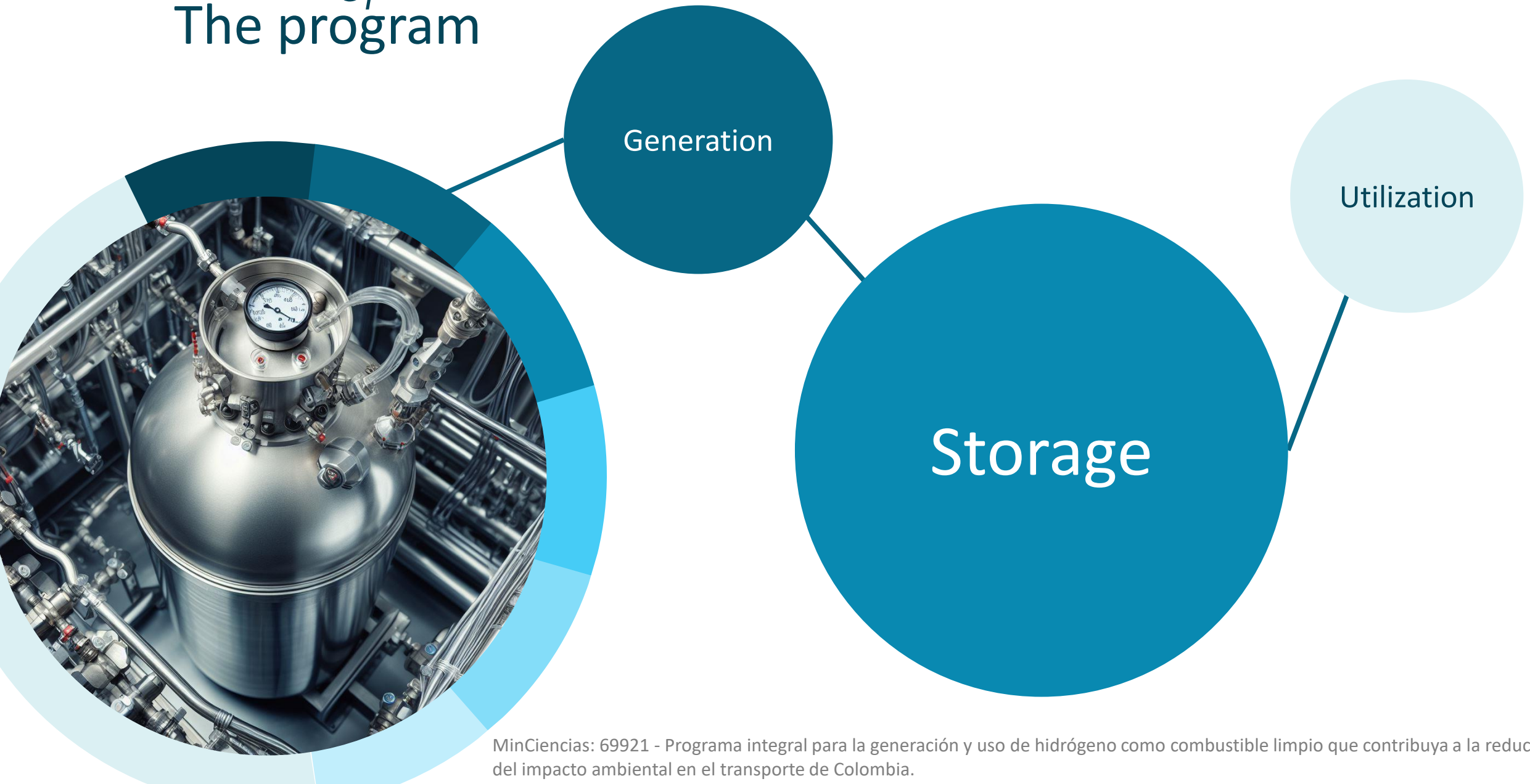
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**Centro de investigación
innovación y desarrollo de
Materiales - CIDEMAT**
Universidad de Antioquia
2023

Overview *Of* The program

UdeA



MinCiencias: 69921 - Programa integral para la generación y uso de hidrógeno como combustible limpio que contribuya a la reducción del impacto ambiental en el transporte de Colombia.

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🎯 Objective

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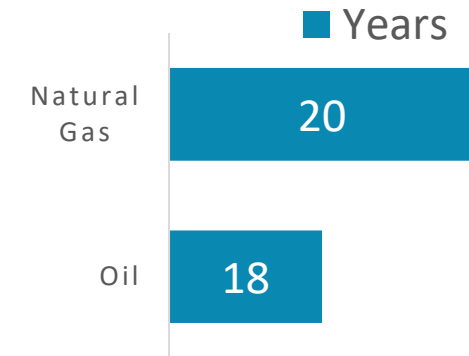
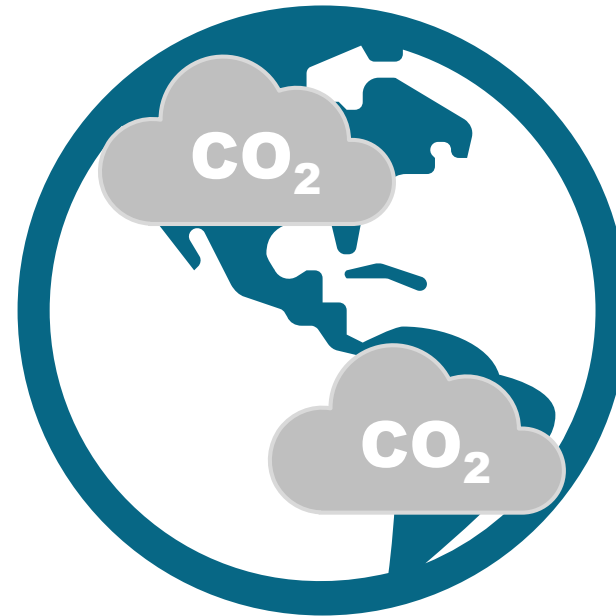
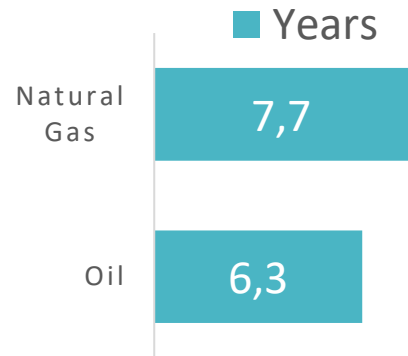
👣 Future Steps

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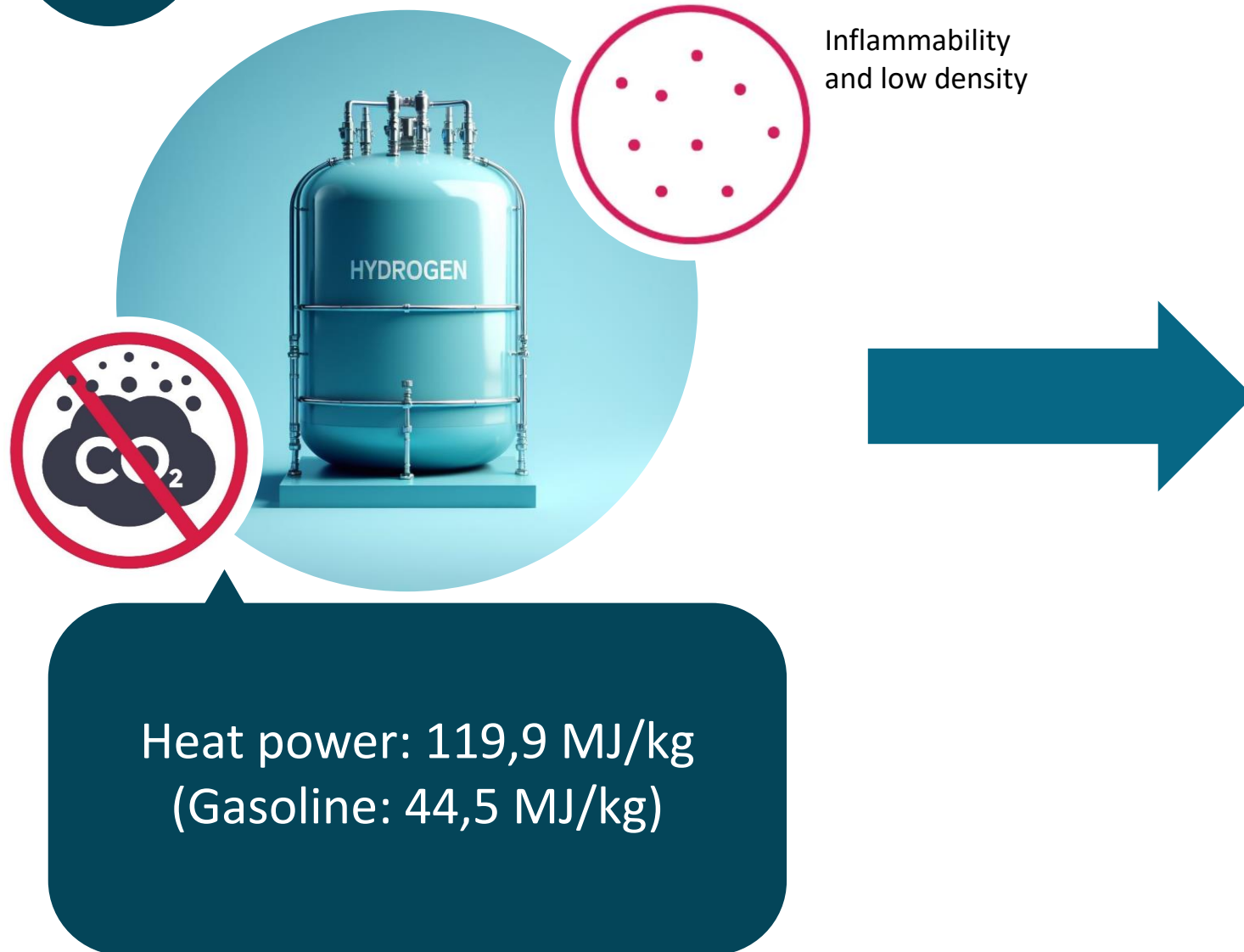
Introduction | fossil fuels



Sustainable alternatives are required



Introduction | energy transition



Storage is limiting its practical applications.



Introduction | hydrogen storage

Solutions for H₂ storage

Gases

Liquids (LHCs)

Solids



Lower risk in transportation and higher gravimetric capacities.



Magnesium hydrides (MgH₂)

Lightweight, reversible and abundant on Earth's crust

Theoretical storage capacity (7.6 wt.%)

Sluggish kinetics and thermodynamic stability

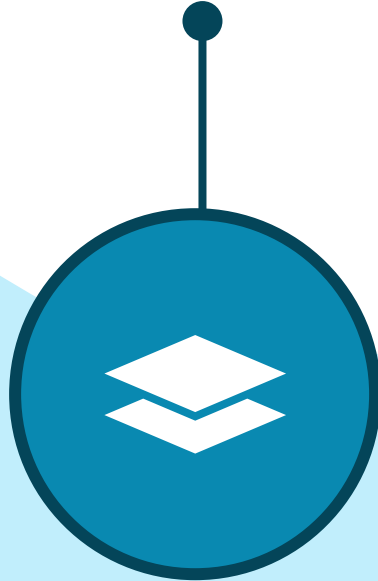
Sorption /
Desorption
>300 °C



Introduction | Mg modifications



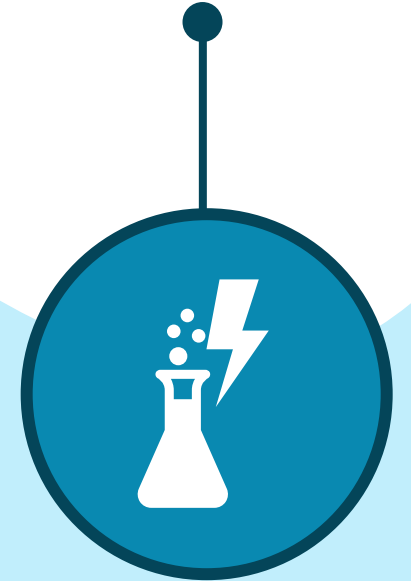
Nanoparticles or
nanostructures



Morphological
modifications



Alloying
elements and
additives



Metal-oxide
catalysts

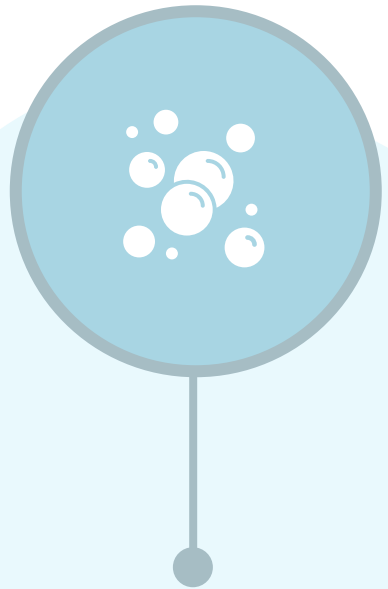


Introduction | Mg modifications

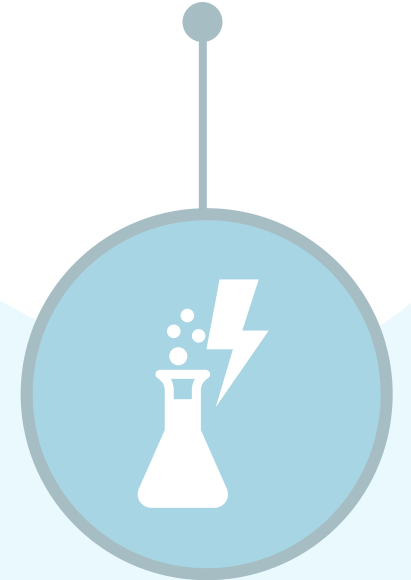
Improve kinetics or thermodynamics
But decrease storage capacity

An approach that involves various modifications is required

Metal-oxide
catalysts



Nanoparticles or
nanostructures



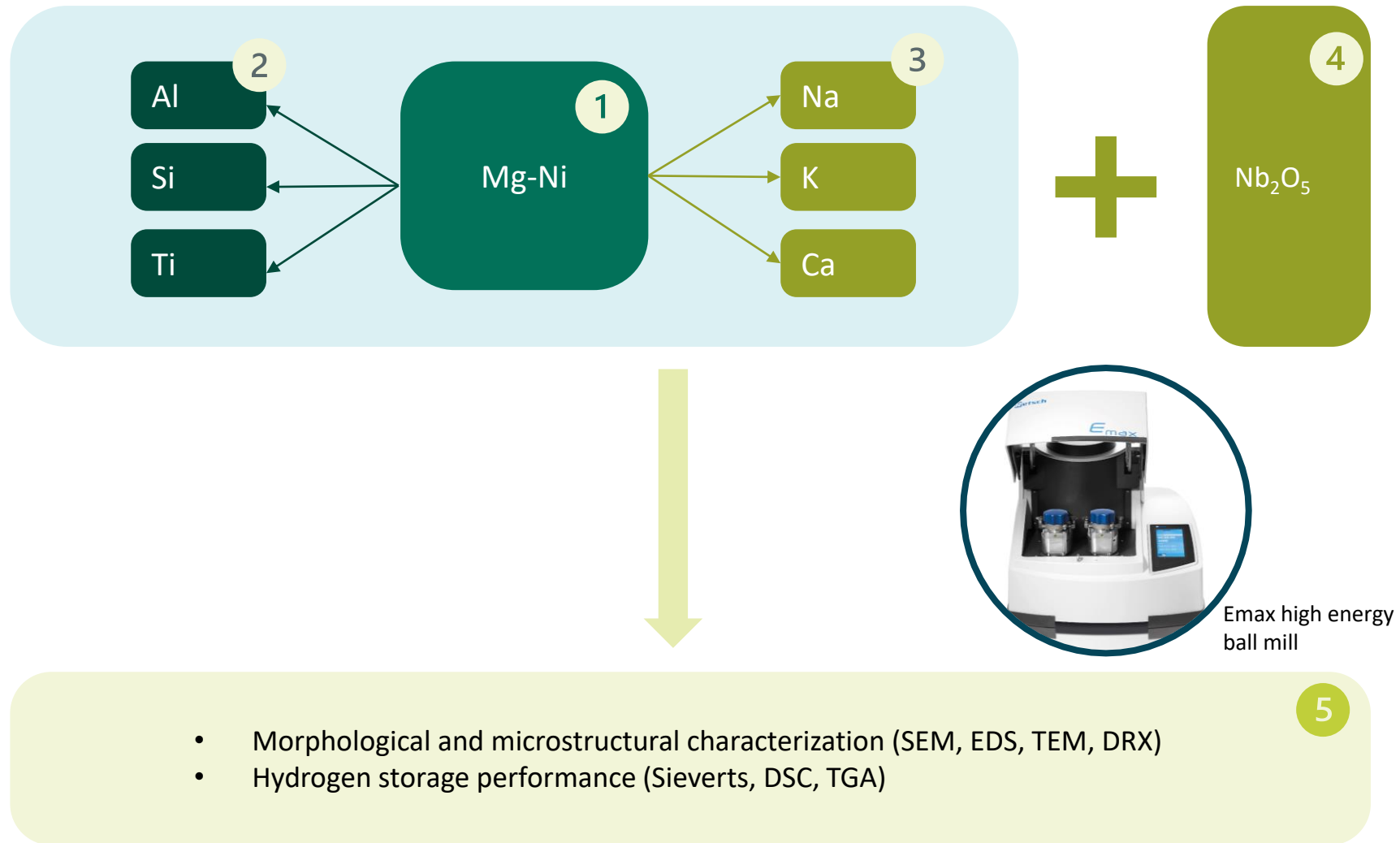
Alloying
elements and
additives



To develop a complex nanostructured Mg-Ni based hydride through high energy ball milling for hydrogen storage applications.



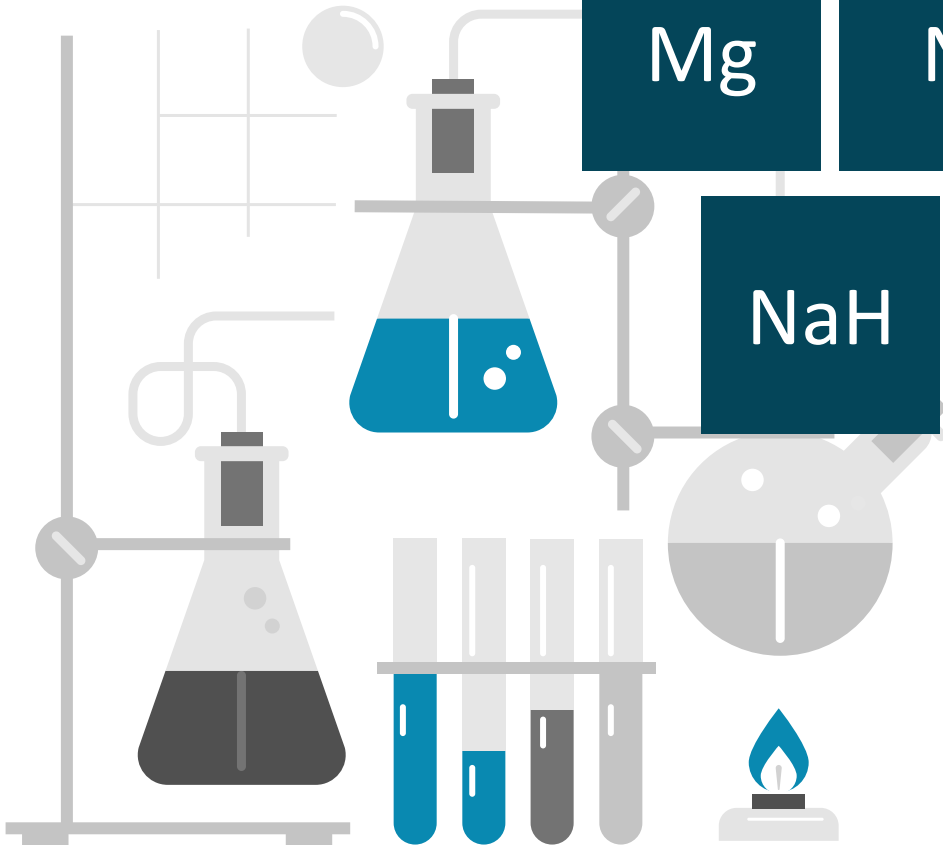
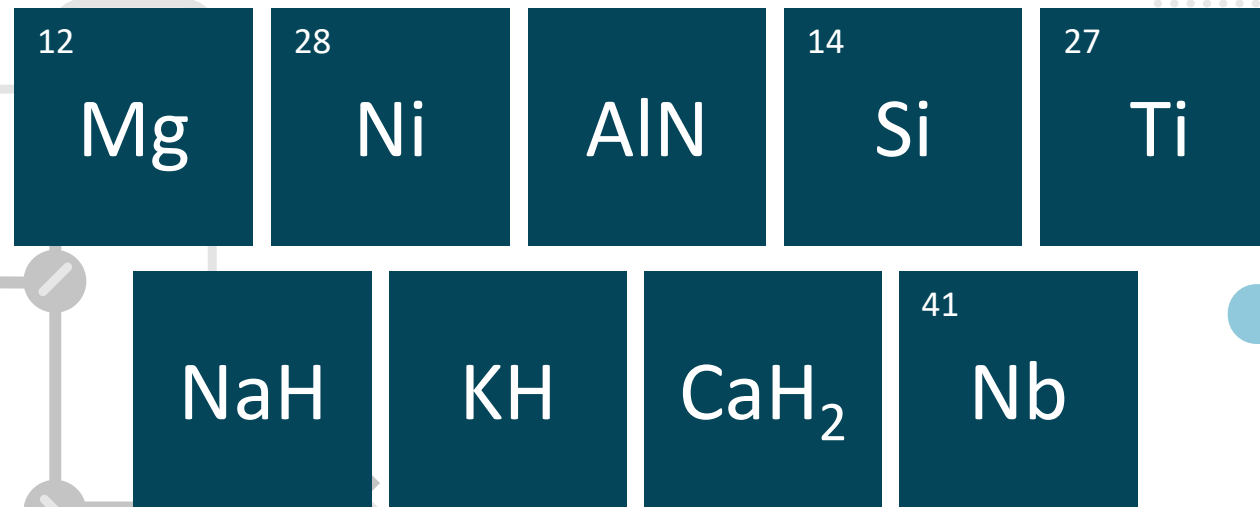
Methodology





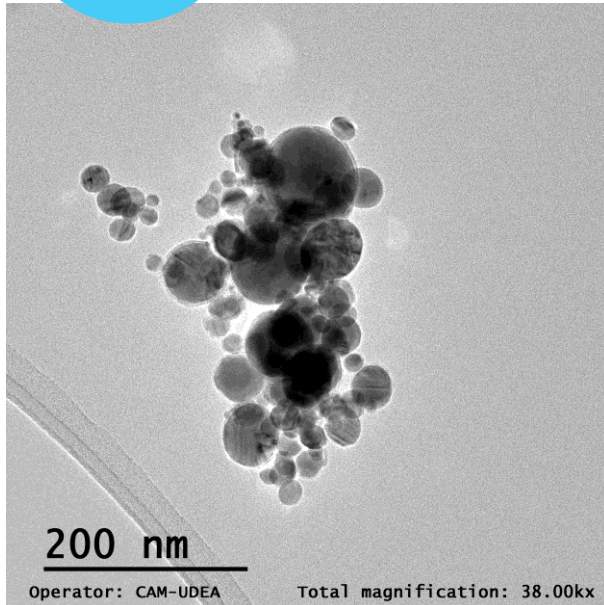
Results | Starting materials

UdeA





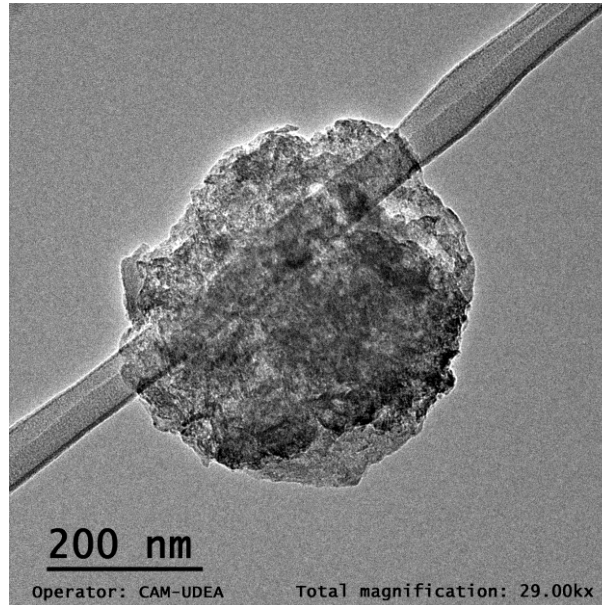
Results | Starting materials



28

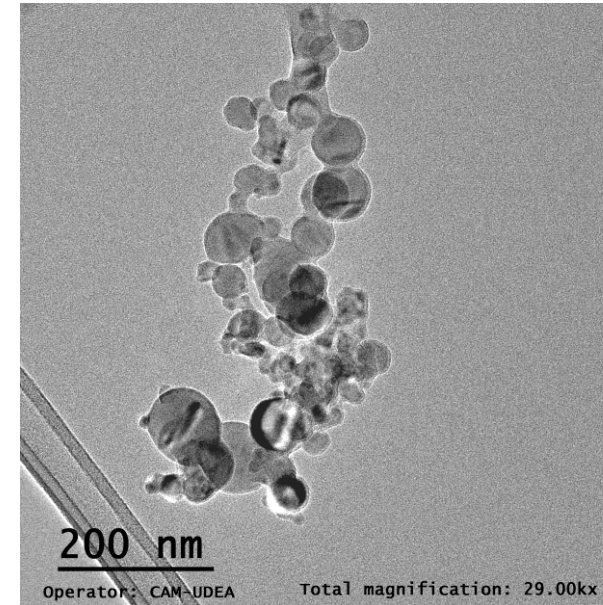
Ni

$28,3 \pm 16,1$ nm



AlN

$362,5 \pm 184,6$ nm

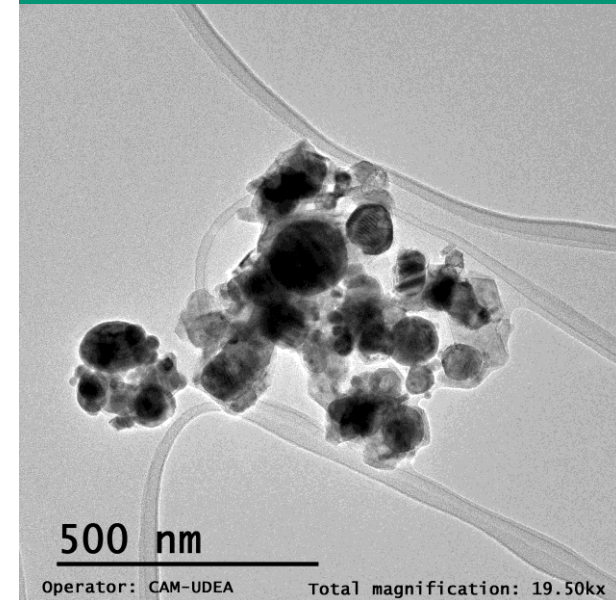


14

Si

$53,9 \pm 22,2$ nm

Controlled oxidation in oxygen rich atmosphere



41

Nb

$59,5 \pm 36,5$ nm

NaH

KH

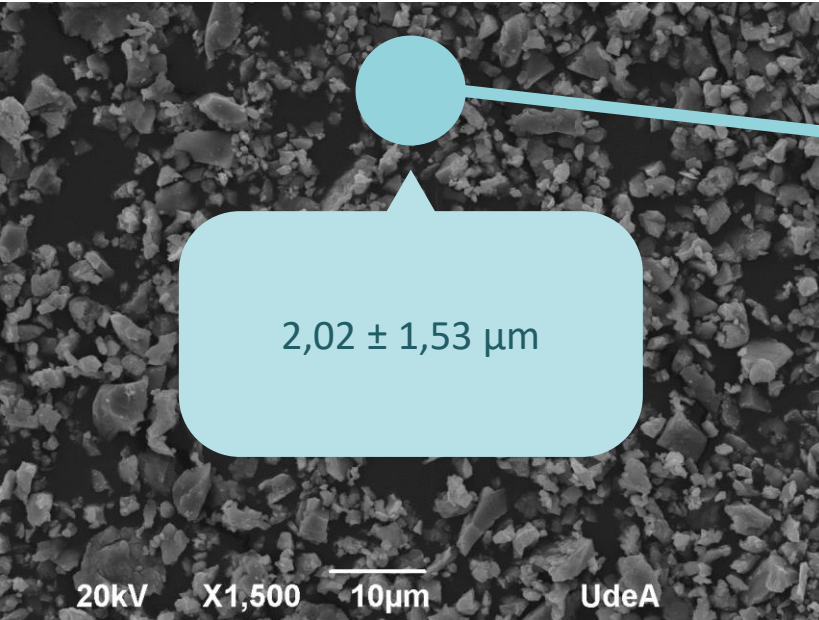
CaH₂

No NPs

10 mm, 6h

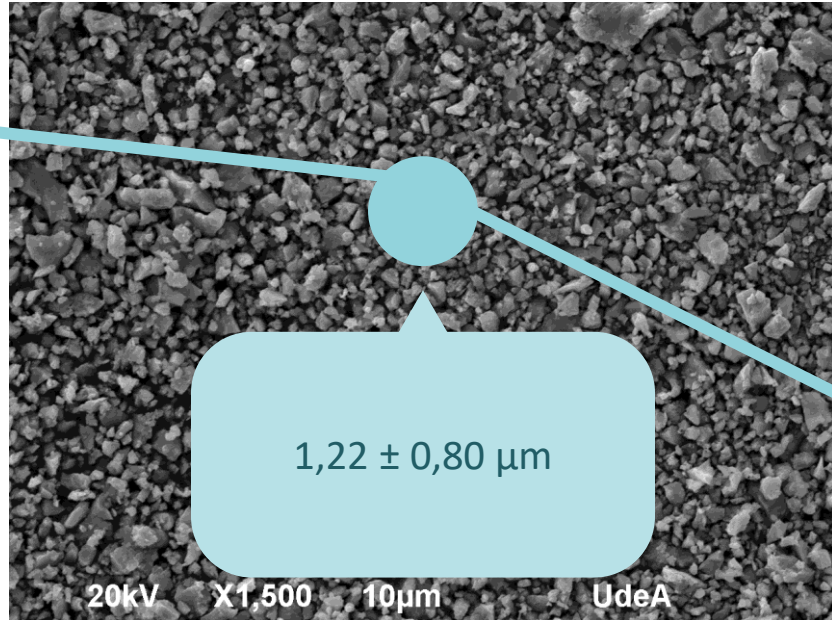
3 mm, 6h

1 mm, 6h



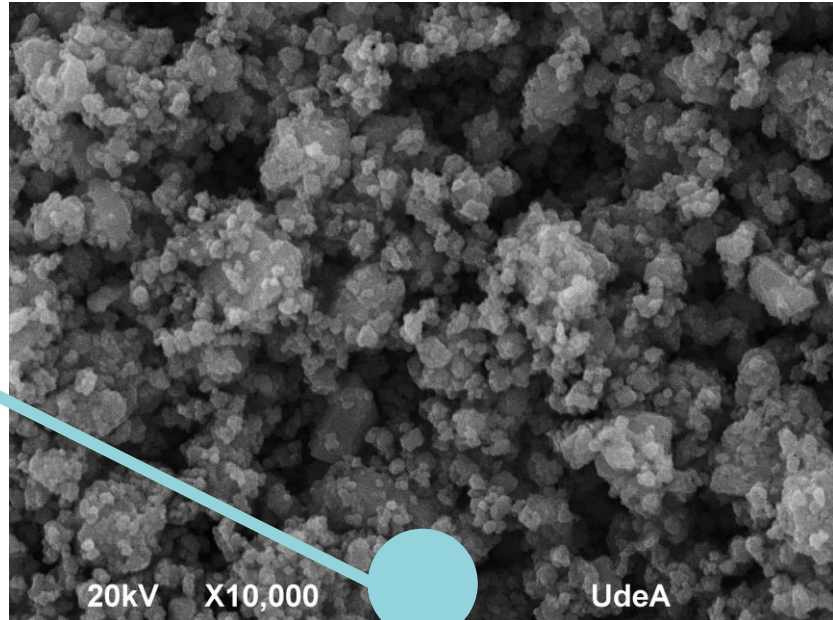
$2,02 \pm 1,53 \mu\text{m}$

20kV X1,500 10μm UdeA



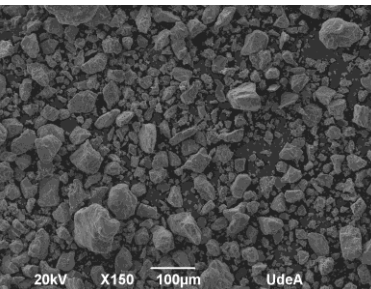
$1,22 \pm 0,80 \mu\text{m}$

20kV X1,500 10μm UdeA



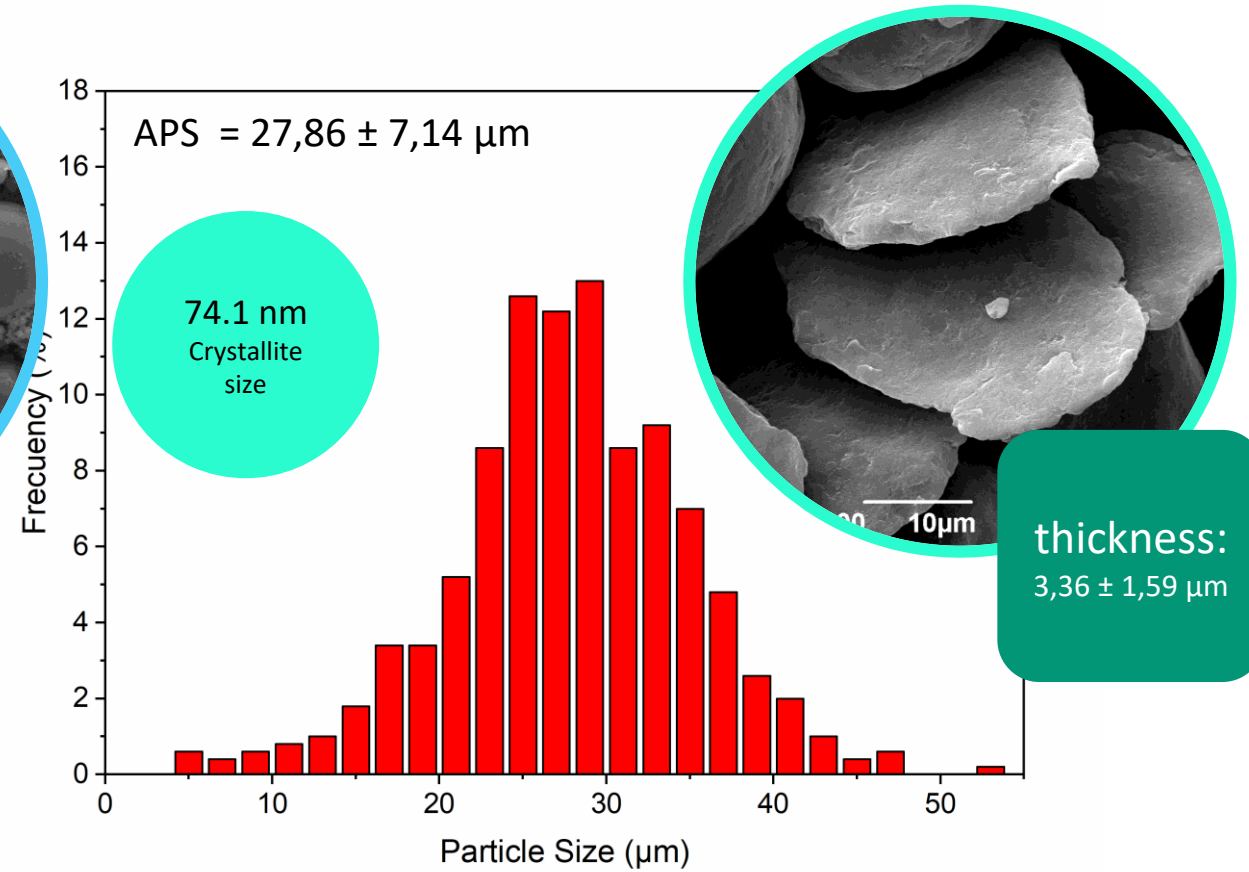
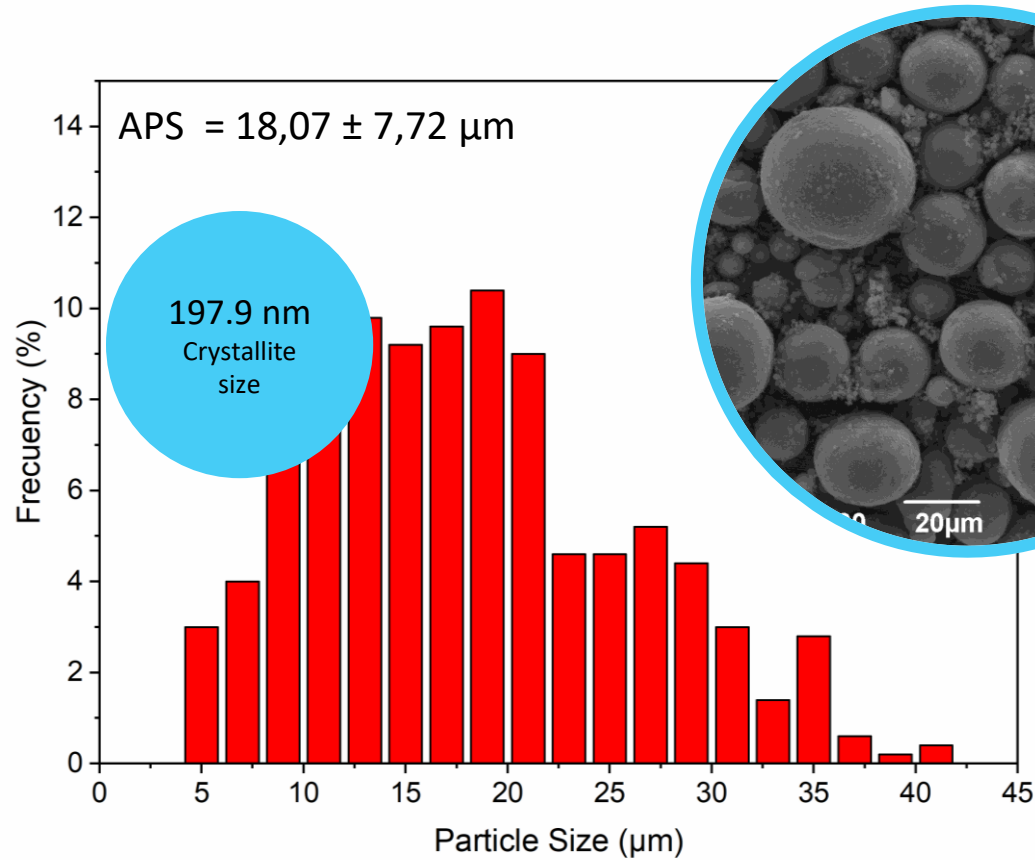
$171 \pm 122 \text{ nm}$

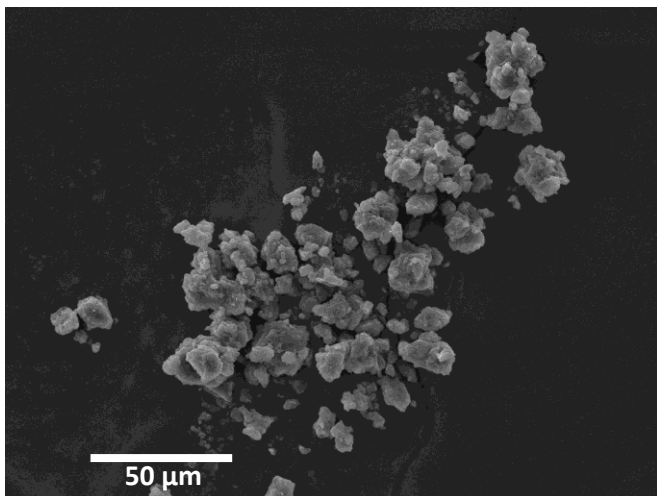
20kV X10,000 UdeA



Starting material

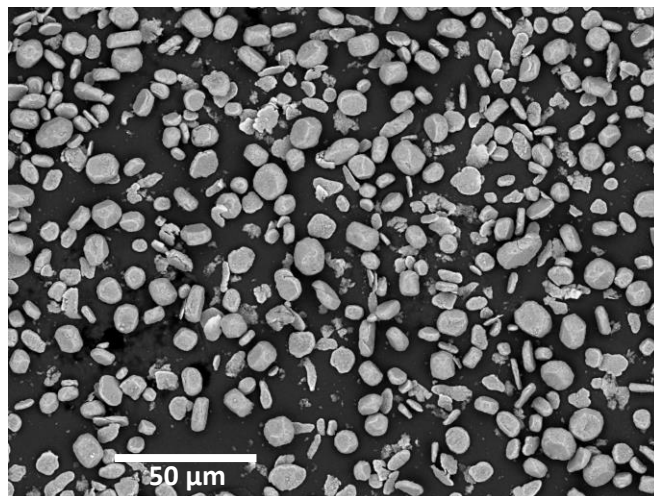
20kV X150 100μm UdeA





Sonochemistry

- Ultrasound probe
- $\text{MgCl}_2 + \text{LiH}$



Cryo-milling

- Milling with LN_2
- cp-Mg

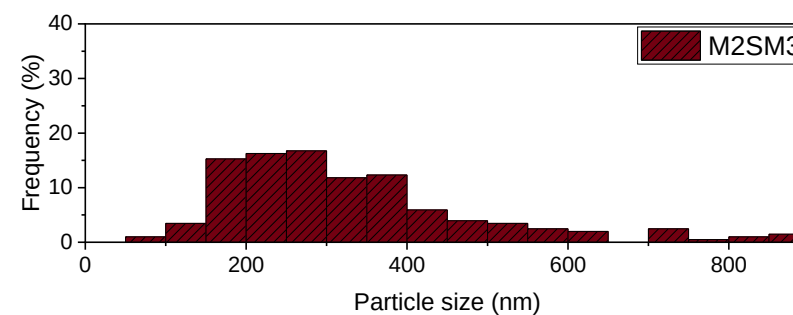
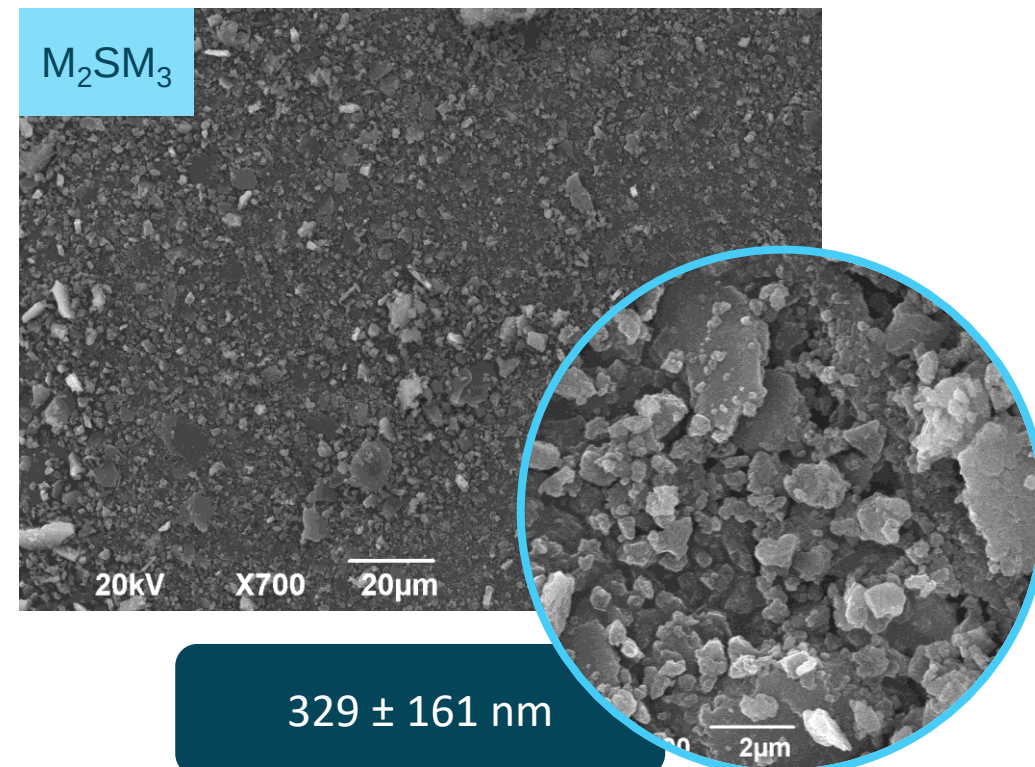
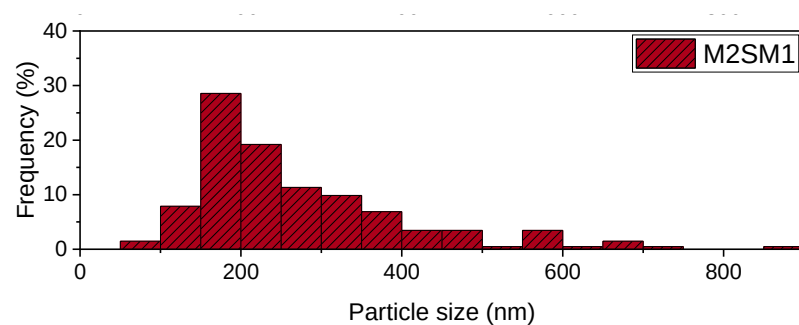
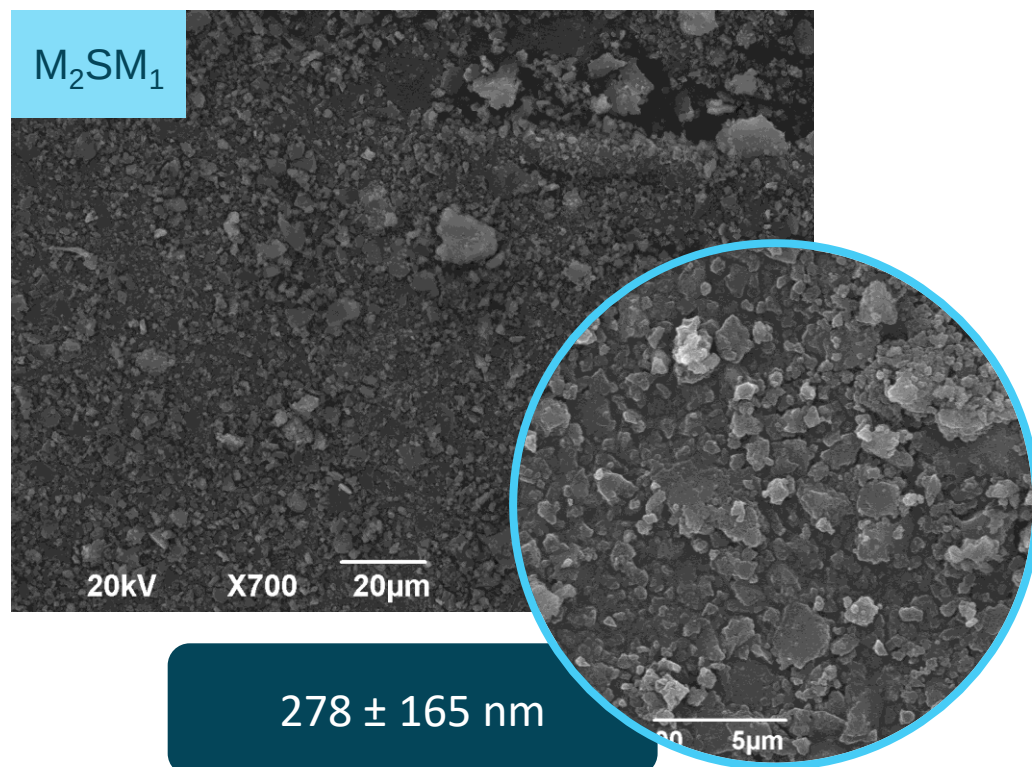
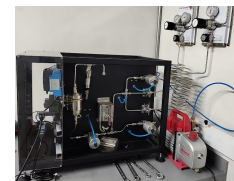
Sieverts type apparatus



Three-step process

- Emax - Sieverts
- cp-Mg

Results | Three-step process

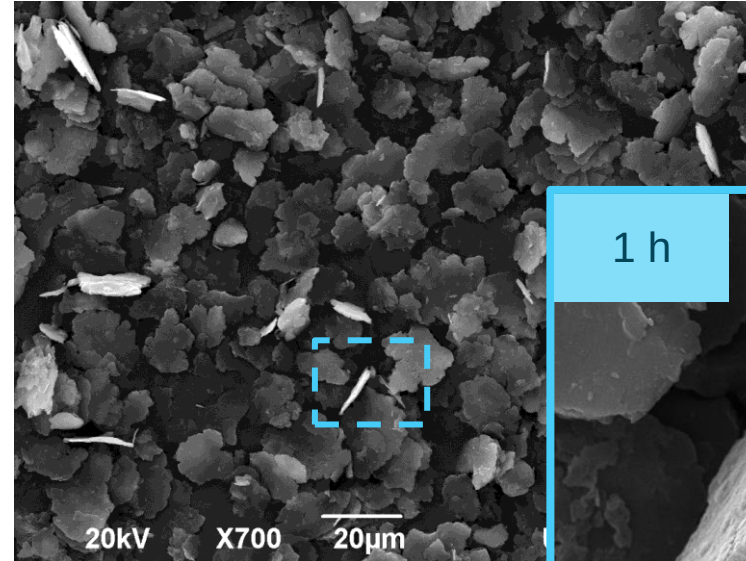


- Mg cp
- ZrO₂ Balls and Jar
- 1400 rpm

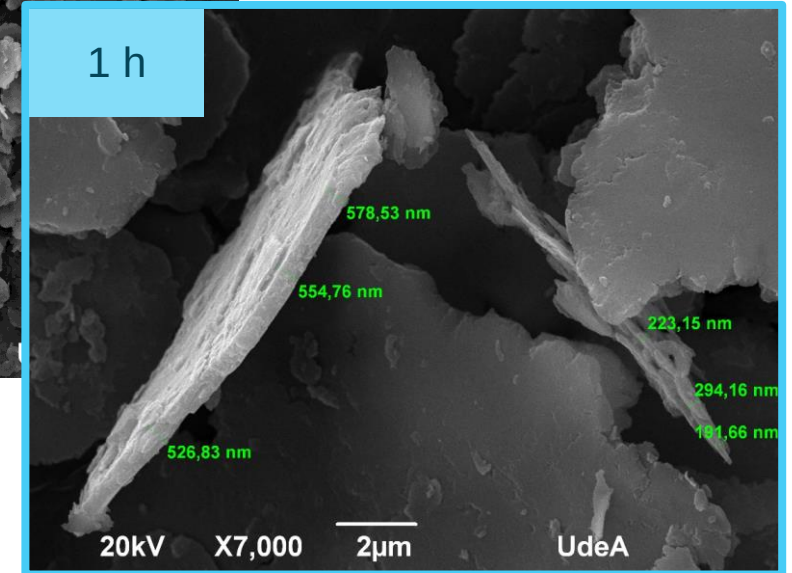
1st milling

2nd milling

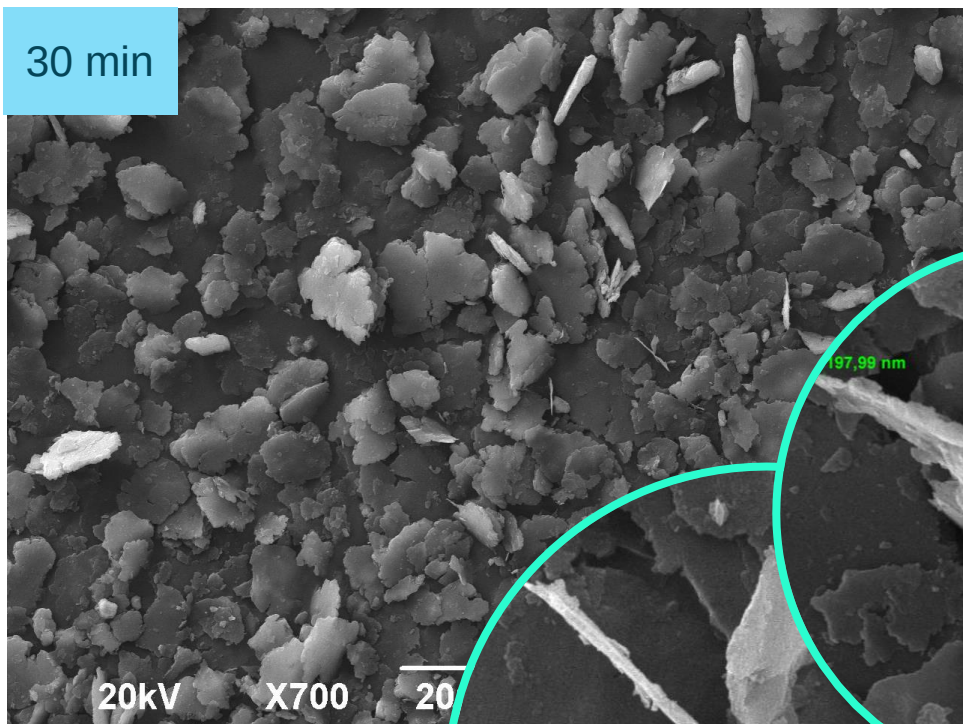
- Mg from 1st step
- ZrO₂ Balls and Jar
- 1400 rpm



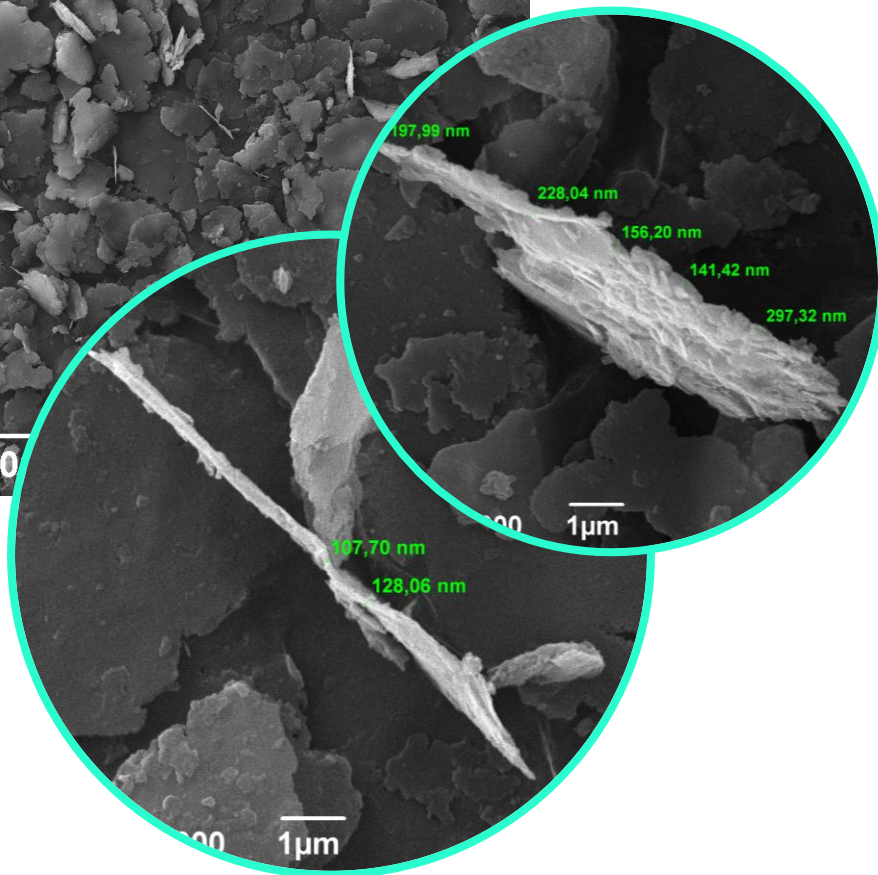
$10,78 \pm 4,47 \mu\text{m}$



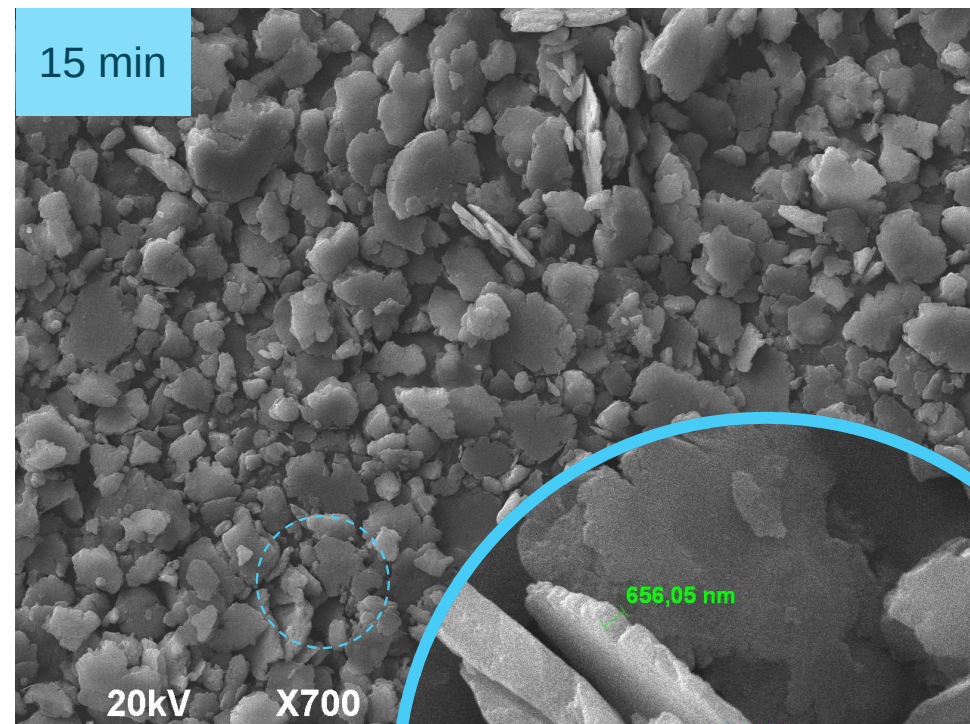
30 min



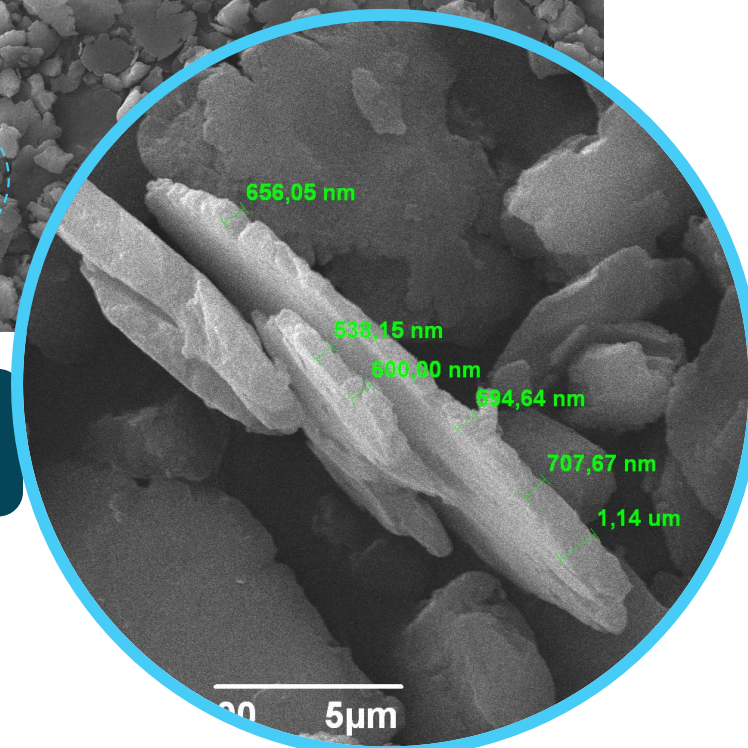
$9,38 \pm 3,55 \mu\text{m}$



15 min

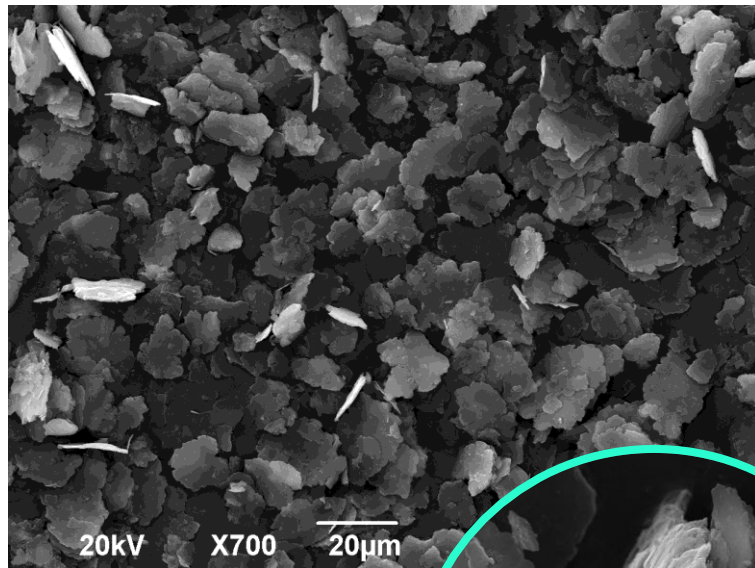


$8,91 \pm 4,37 \mu\text{m}$

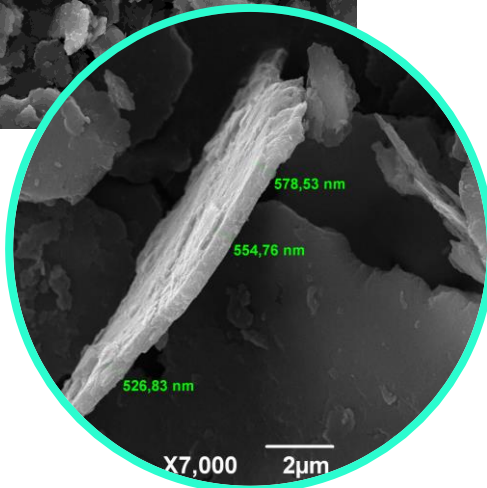




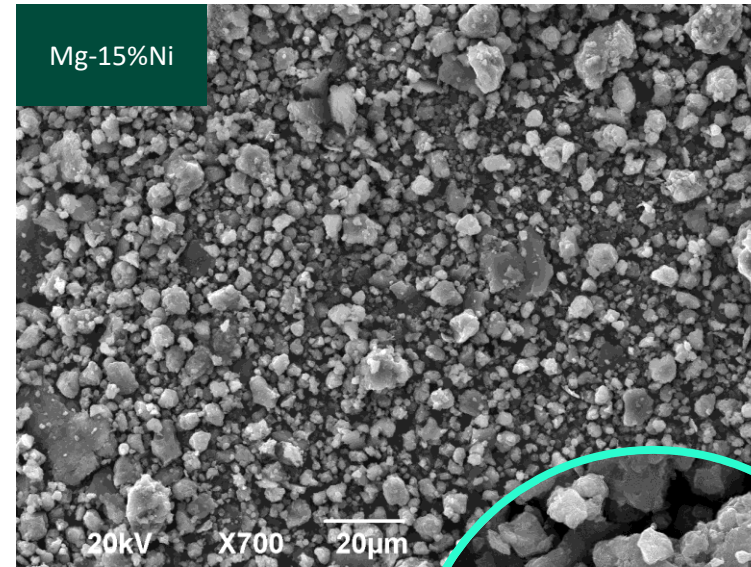
Results | Mg-15%Ni (flakes)



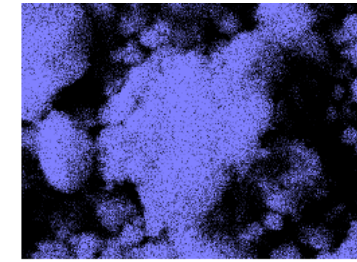
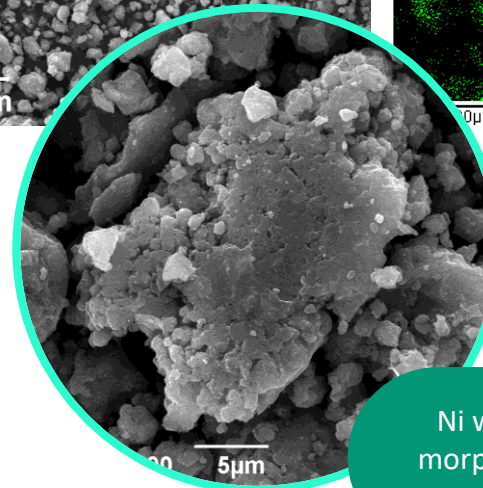
$10,78 \pm 4,47 \mu\text{m}$



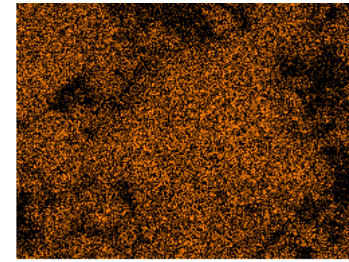
+Ni



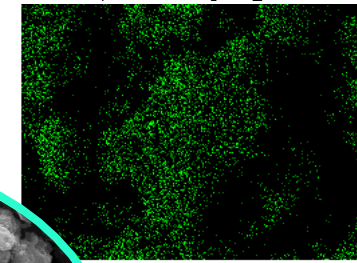
$2,46 \pm 1,26 \mu\text{m}$



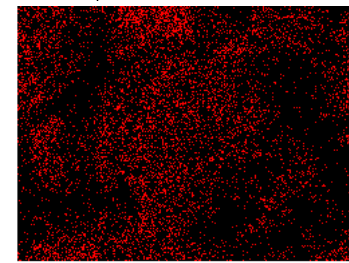
20µm Mg Ka1_2



20µm Ni Ka1



20µm O Ka1

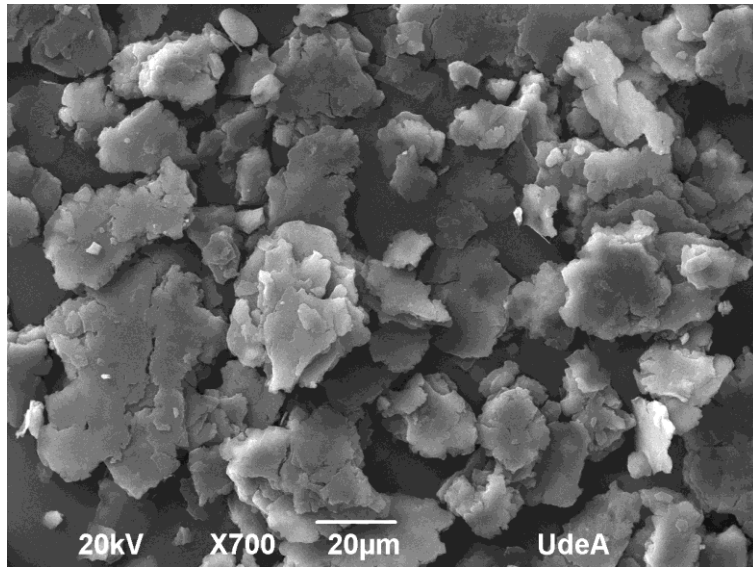


20µm C Ka1_2

Ni was dispersed but morphology underwent alteration

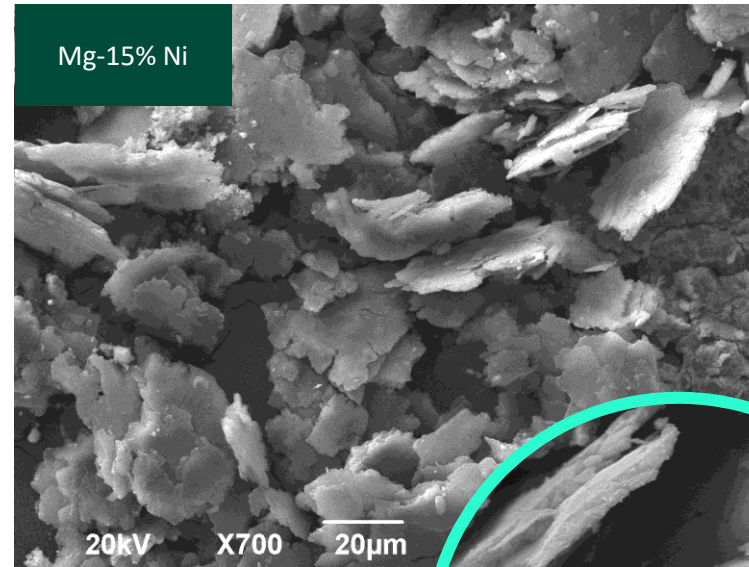


Results | Blending vs. milling

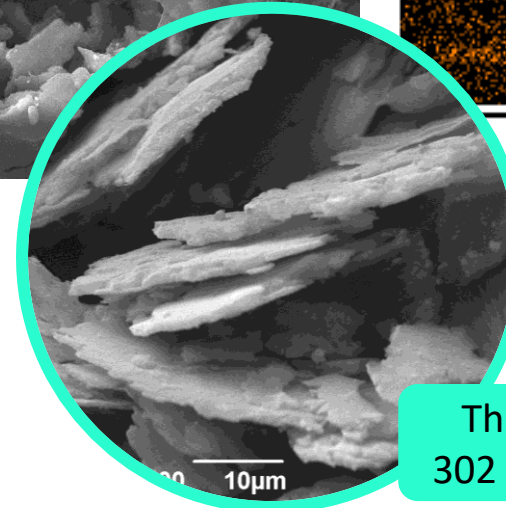


$14,42 \pm 5,94 \mu\text{m}$

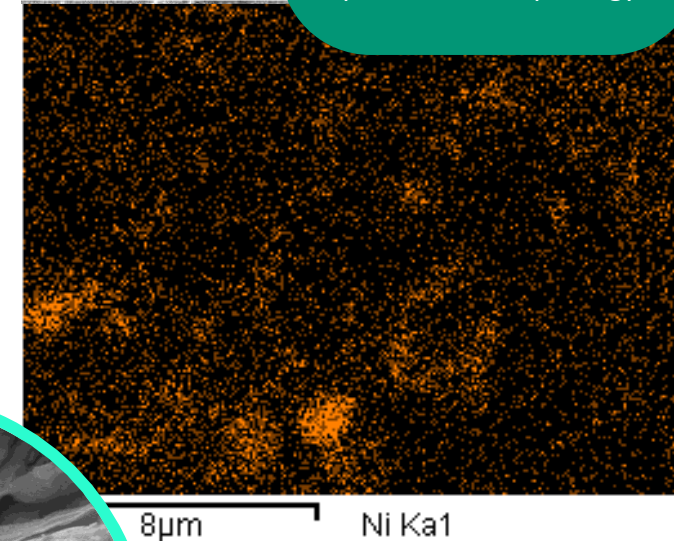
+Ni



$18,92 \pm 7,59 \mu\text{m}$



Thickness:
 $302 \pm 127 \text{ nm}$

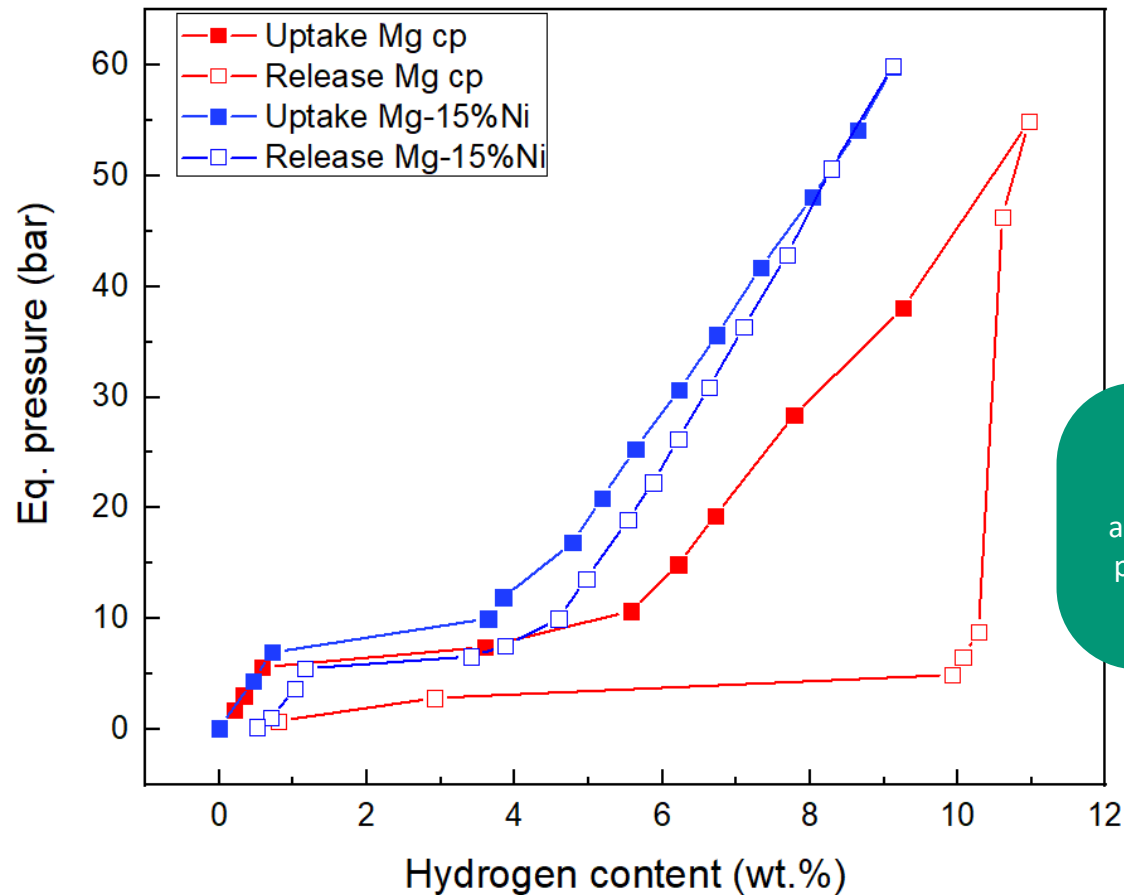


Dispersed Ni and
preserved morphology

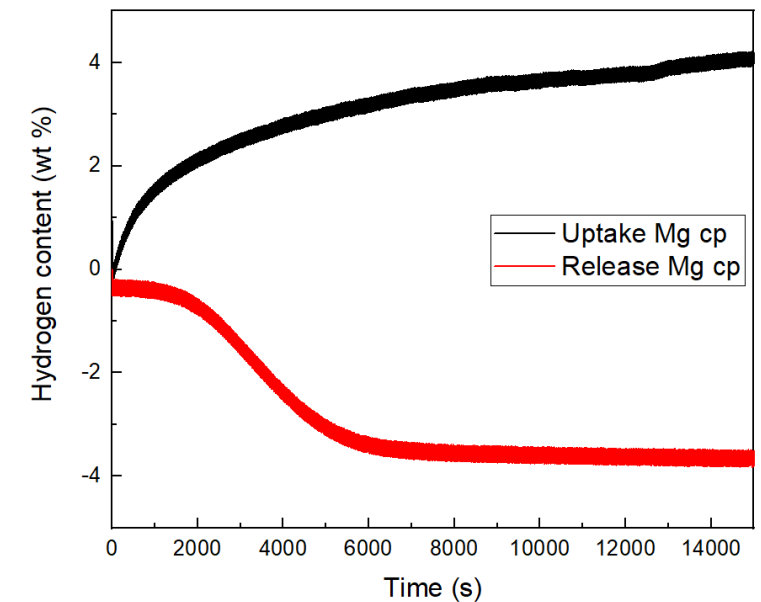
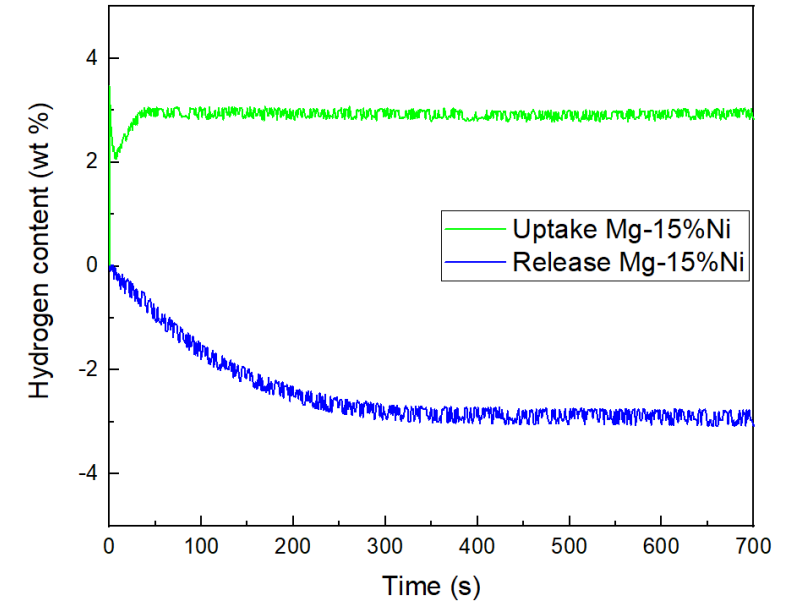


Results | Preliminary hydrogen tests

350°C

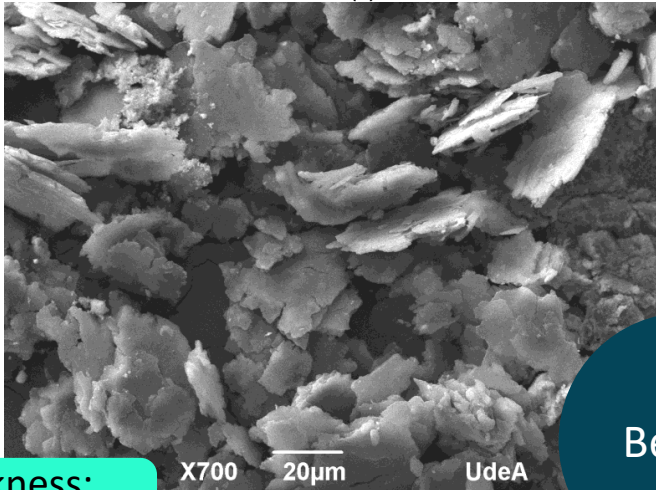
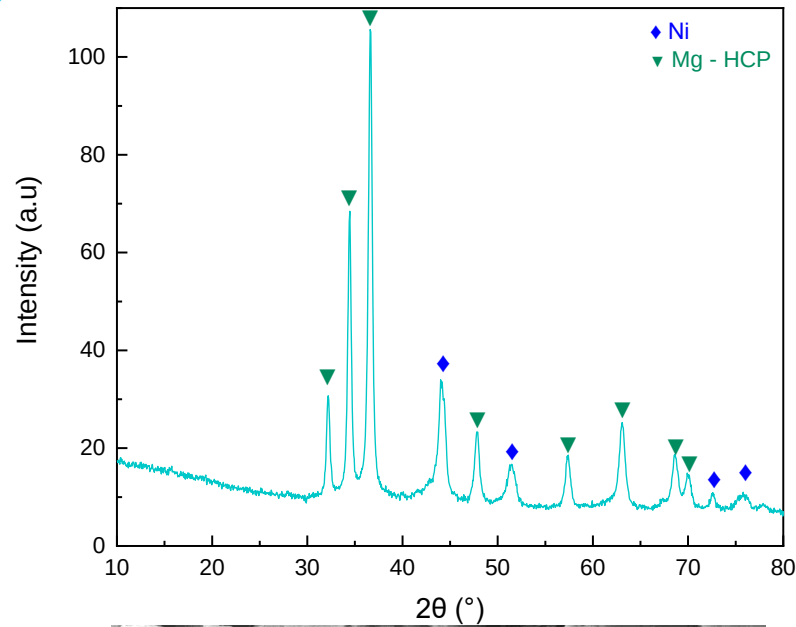


Sieverts type apparatus requires proper calibration





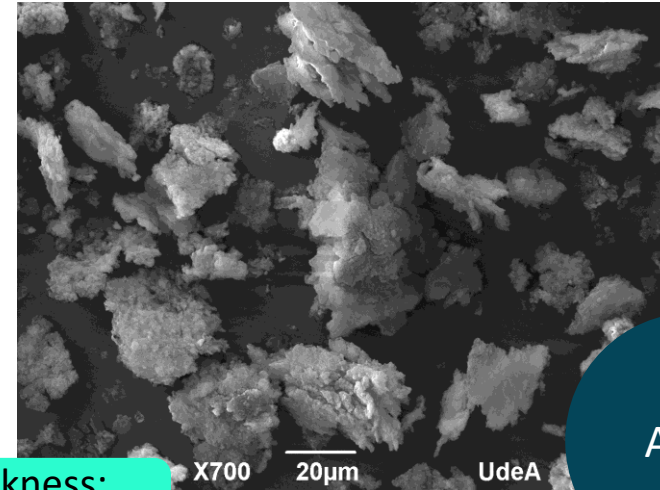
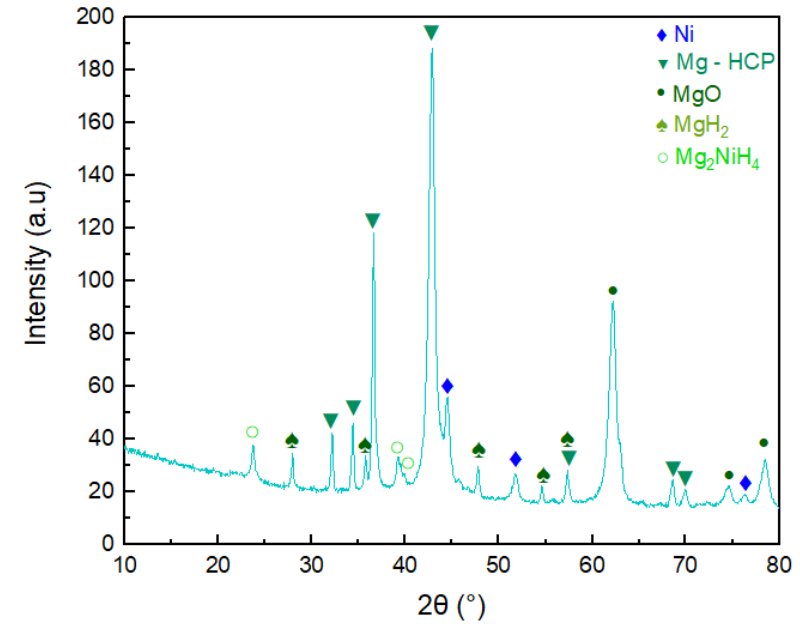
Results | Preliminary hydrogen tests



Thickness:
 $302 \pm 127 \text{ nm}$

Before

H₂
tests



Thickness:
 $426 \pm 111 \text{ nm}$

After



Future steps

Preliminar

Sieverts
calibration

Flakes - Ni system
evaluation

Materials

Flakes - Si, Ti, AlN
blending

Flakes - NaH, KH,
CaH₂ blending

Nb₂O₅ addition

Characterization

Storage
performance
(Sieverts)

XRD, SEM-EDS

Bibliography



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- U.S. Department of Energy. (2023). *Hydrogen Storage*. Energy.gov. Retrieved October 13, 2023, from <https://www.energy.gov/eere/fuelcells/hydrogen-storage>

Thanks *for* your attention



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